

CRYSTAL CAVE

Wisconsin's Longest Show Cave Educational Program

Cave Tour Lesson Plan Grades 5-6

Objectives:

At the end of this program, the student should be able to:

- Define cave related vocabulary
- Explain what a cave is
- Tell how a cave forms in general terms
- Name at least 4 cave formations
- Express a connection between what is above ground and what is below ground

Wisconsin DPI Standards

Science:

A.8.1, C.8.1, D.8.2, D.8.3, D.8.4,
D.8.5, D.8.6, D.8.7, D.8.8, E.8.1,
E.8.2, E.8.3, E.8.4, E.8.5, E.8.6

Social Studies:

A.8.1, A.8.2, A.8.4, A.8.

Minnesota Academic Standards

Science:

5.1.1.2.1, 5.3.1.2.1, 5.3.1.2.2, 6.2.1.2.1

Activities:

Times are approximate and specific reinforcement activities will vary based on the needs of each individual group.

10 minutes: The visual presentation provides the history and discovery of the cave, definition of a cave, formation of sedimentary rocks, how caves form, fossils in the cave, different types of cave formations, the type and hibernation of bats, and the ecology of caves.

60 minutes: The Cave Tour fosters a connection between previously discussed cavern features and formations with the experience of the actual cave environment. A knowledgeable guide shows the group through 11 rooms on three levels.

Pre-teach Vocabulary:

A glossary of terms is provided for your convenience.

Acid - A substance that produces ions when it is dissolved in water. Acids can breakdown (dissolve) rocks and minerals.

Calcite - A mineral composed of calcium carbonate. Most cave formations are made of calcite.

Cave - A hole in rock that was made by nature and is large enough for a person to fit into.

Column - A formation which is formed when stalagmites meet overhanging stalactites. Water flowing down the sides of the column gradually enlarges it by adding layers of flowstone to the surface.

Crystal - A solid whose atoms are arranged in an orderly, repeating, three-dimensional pattern. All minerals, such as calcite, are composed of crystals.

Dissolve - To breakdown a substance into smaller more dilute particles.

Environment - All the external conditions, which surround a living thing.

Erosion - The set of processes by which materials are removed or transported by wind, water, ice or gravity

Flowstone – Sheets of calcite deposited in a cave as mineral rich water flows over the wall or floor.

Fossil - Any remains or traces of animals or plants that lived in the past. These can include bones, tracks, casts or imprints.

Geology – Scientific study of the earth and the materials that form it.

Limestone - A carbonate-rich sedimentary rock which usually forms from layers of the remains of marine life and other marine sediments.

Mineral - A naturally occurring, solid element or compound, with a definite composition and a regular internal crystal structure.

Rock - A solid, cohesive aggregate of one or more minerals or mineral materials.

Sinkhole - A circular depression on the surface formed by ground collapsing into a cavity below.

Speleology - The scientific study of caves.

Speleothem - A general term for any mineral deposit or formation found within a cave, such as a stalactite or stalagmite.

Stalactite - A formation which develops when water deposits minerals in successive rings downward from the ceiling of a cave.

Stalagmite - A formation which builds upward from a cave floor as the result of water dripping from above. They are usually located beneath a stalactite.

Weathering - The breakdown of earth materials by some force of nature such as wind, water or gravity.

Learning Extension:

Try this activity after your visit to reinforce important concepts.

You will need:

- ~White vinegar
- ~Shallow pan
- ~Eye droppers
- ~Limestone sample
- ~Other rock samples

Directions:

1. It may be helpful to break into partners or teams of 3.
2. Place rock samples in the shallow pan.
3. Use eye droppers to coat each rock with the vinegar
4. Make and record observations.

Questions: What happened to each rock as the vinegar was added? Why do you think vinegar reacted that way? What do you think would happen if you coated the rocks with just plain water? Would the reaction be the same or different? Why?

Discuss: You have just performed what is called a bubble test. Limestone will dissolve in acid. Vinegar is an acid that is strong compared to carbonic acid. Carbonic acid forms when rain picks up carbon dioxide from the atmosphere and soil. As this weak acid moves through limestone, the rock is slowly dissolved leaving behind cavities. The dissolved limestone is then re-deposited into the cave as speleothems made of calcite.